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Chemistry 11 Final Exam Study Guide Chemistry 11 - Final Exam Study Guide Page 18. An enthalpy diagram for an exothermicreaction: So, to summarize, in an exothermicreaction: ΔH is negative Products are lowerthan Reactants on the Enthalpy Diagram Heat is released to the surroundings. In an endothermicreaction: Heat is absorbedfrom the surroundings and converted into enthalpy.

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(commonly called the General Chemistry Study Guide) This guide includes 201 pages of information and over 600 problems separated into first-term and second-term general chemistry material. Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students.

Student Study Materials | ACS Exams

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