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Chemistry Chapter 9 Vocab. chemical equation. chemical reaction. coefficient. combustion reaction. a short, easy way to show a chemical reaction, using symbols a.... a process in which one or more substances are changed into dif.... number in front of a chemical formula in an equation that indi....

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Chapter 9 Carbon Chemistry Summary 9.1 Carbon Compounds • An organic compound contains carbon and hydrogen, often combined with a few other elements such as oxygen and nitrogen. • Carbon has four valence electrons and can bond in several ways with other atoms. Diamond, graphite, and fullerenes are forms of carbon.

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Section 9.1 Assessment page 288 7. Explain why it is important that a chemical equation be balanced. Because mass is neither created nor destroyed in chemical reactions, the numbers of atoms of all elements must be equal on both sides of the reaction arrow. 8. List three types of physical evidence that indi-cate a chemical reaction has occurred.

Chemical ReactionsChemical Reactions

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Chapter Assessment Complete the table. Principal Quantum Number, n Types of Orbitals Number of Orbitals Related to Principal Energy Level Write the orbital diagram and complete electron configuration for each atom. 9. nitrogen 10. fluorine is 11. sodium 26 Chemistry: Matter and Change Chapter 5