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quantity that has both a number and unit. an expression of numbers in the form $m \times 10^n$ where m is equal.... a measure of how close a measurement comes to the actual or tr....

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Section 3: Scientific Notation. The study of chemistry can involve numbers that are very large. It can also involve numbers that are very small. Writing out

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such numbers and using them in their long form is problematic, because we would spend far too much time writing zeroes, and we would probably make a lot of mistakes!

Chapter 1: Measurements in Chemistry - Chemistry

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Chapter 3 - Scientific Measurement - 3.1 Using and ...

Chapter 3 - Scientific Measurement - 3.1
Using and Expressing Measurements -

3.1 Lesson Check - Page 72: 13 Answer

To evaluate the accuracy of a
measurement, the measured value must

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be compared to the correct value.

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Assignment & Problem Set 5. Express the following numbers in proper scientific notation. a. 0.00567 b. 543000 c. 56.7×10^{-4} d. 0.085×10^7 6. Express the following numbers in decimal notation a. 4.94×10^5 b. 6.062×10^{-5} 7. Solve each problem and express each answer in scientific notation. a.

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Scientific Measurement Chapters 3 Assignment & Problem Set

after reading Lesson 3.1, answer the following questions. Scientific notation 1. Why are numbers used in chemistry often expressed in scientific notation? 2. Circle the letter of each sentence that is true about numbers expressed in scientific notation. a. A number

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expressed in scientific notation is written as the product of a coefficient

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