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Chemistry I Chapter 10 Give the approximate bond angle for a molecule with a trigonal planar shape. Give the approximate bond angle for a molecule with a tetrahedral shape. Determine the electron geometry (eg) and molecular geometry (mg) of BCl₃. eg=trigonal planar, mg=trigonal planar Determine the electron geometry (eg) and molecular geometry (mg) of CO₃²⁻. eg=trigonal planar, mg=trigonal

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Chemistry Chapter 10 & 14 Vocab. Thermodynamics. Energy. Potential energy. Kinetic energy. the study of energy. the ability to do work or produce heat. stored energy due to position or composition. energy due to the motion of the object and depends on mass. Thermodynamics. the study of energy. Energy.

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CHAPTER 10 SOLUTIONS MANUAL The MoleThe Mole Solutions Manual Chemistry: Matter and Change • Chapter 10 161 Section 10.1 Measuring Matter page 320–324 Practice Problems pages 323–324 1. Zinc (Zn) is used to form a corrosion-inhibiting surface on galvanized steel.

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NCERT Solutions CBSE Sample Papers Chemistry Class 12 Chemistry. NCERT IN TEXT QUESTIONS. 10.1 Write structures of the following compounds: (i) 2-Chloro-3-methylpentane. (ii) 1-Chloro-4-ethylcyclohexane. (iii) 4-tert. Butyl-3-iodoheptane. (iv) 1,4-Dibromobut-2-ene. (v) 1-Bromo-4-sec. butyl-2-methylbenzene.

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