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17.1: Rates of reactions and rate laws. Chemical change is guided and driven by energetics, but the actual route it takes and the speed with which it occurs is the subject of "dynamics". Dynamics is itself divided into two general areas: kinetics, which deals with the rate of change and is the subject of this lesson.

17: Chemical Kinetics and Dynamics - Chemistry LibreTexts

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Chemical kinetics - Theories of reaction rates | Britannica

Chemical Kinetics and Reaction Dynamics. DIVThis text teaches the principles underlying modern chemical kinetics in a clear, direct fashion, using several examples to enhance basic understanding. Solutions to selected problems. 2001 edition. /div.

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Kinetics •Studies the rate at which a chemical process occurs. •Besides information about the speed at which reactions occur, kinetics also sheds light on the reaction mechanism (exactly how the reaction occurs).

Chemical Kinetics - Duke University

Reaction dynamics is a field within physical chemistry, studying why chemical reactions occur, how to predict their behavior, and how to control them. It is closely related to chemical kinetics, but is

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concerned with individual chemical events on atomic length scales and over very brief time periods.

Reaction dynamics - Wikipedia

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Chemical Kinetics and Reaction Dynamics

Chemical kinetics is the study of chemical processes and rates of reactions. This includes the analysis of conditions that affect speed of a chemical reaction, understanding reaction mechanisms and transition states, and forming mathematical models to predict and describe a chemical reaction.

What Is Chemical Kinetics?

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study of rates of chemical processes Chemical kinetics, also known as reaction kinetics, is the branch of physical chemistry that is concerned with understanding the rates of chemical reactions. It is to be contrasted with thermodynamics, which deals with the direction in which a process occurs but in itself tells nothing about its rate.

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