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## Chapter 16

### Thermal Energy

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a difference in

temperature.

Temperature. a

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measure of how hot or cold an ...

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Chapter 16 Thermal Energy and Heat

Summary 16.1 Thermal Energy and Matter

Heat flows

spontaneously from hot objects to cold

objects. • Heat is the transfer of thermal

energy from one object to another because of

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a temperature

difference.

Temperature is related

to the average kinetic

energy of the particles

in

## **Chapter 16 Thermal Energy and Heat**

Physical Science

Chapter 16: Thermal

Energy and Heat. Heat

is the transfer of

thermal energy from

one object to another

as the result of a

difference in

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And Heat produces heat.

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## **Physical Science**

### **Chapter 16: Thermal Energy and Heat ...**

It states that thermal energy can flow from colder objects to hotter objects only if work is done on the system.

Third law of thermodynamics It states that absolute zero cannot be reached.



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Summary 16.1 Thermal  
Energy and Matter

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objects. • Heat is the

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energy from one object to another because of a temperature difference.

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With ...**

is a measure of how  
hot or cold an object is  
compared to a ref....

Increase in volume of

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Thermal Energy

material when its temperature increases.

Conduction, is the transfer of thermal energy through touching with no ove....

Heat Engine. is any device that converts heat into work. 36

Terms. etakp. Chapter 16 Thermal Energy and Heat.

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chapter 16 heat  
thermal energy  
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## Chapter 16

### Chapter 16: Thermal Energy

#### 16.1 – Thermal Energy and Matter

#### Work and Heat

Heat is the transfer of thermal energy from one object to another because of temperature differences

Heat flows spontaneously from hot objects to cold objects. Imagine two glasses with differing amounts of water in them at the same temperature. The glass

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Answers next it is not

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convection, conduction  
and radiation as well as  
the use of insulation.  
Category.

**Chapter 16 -  
Thermal Energy**

16.1 Thermal Energy &  
Matter. Work and Heat.  
Heat -the transfer of  
thermal energy from  
one object to another

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### Thermal Energy

because of a temperature difference. Heat flows from higher temps to lower temps. Temperature is related to the kinetic energy of the particles: particles move around as they heat

### **Chapter 16**

Chapter 16: Thermal Energy and Heat.

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Thermal Energy

Heat: the transfer of thermal energy from one object to another because of a difference in temperature:  
temperature: a measure of how hot or cold an object is compared to a reference point:

### **Quia - Chapter 16: Thermal Energy and Heat**

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Thermal Energy  
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Section 16.1 Thermal Energy and Matter (pages 474–478) This section defines heat and describes how work, temperature, and thermal energy are related to heat

Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained Reading Strategy (page 474)

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Energy and Heat  
Calculating with  
Specific Heat How  
much heat is required  
to raise the  
temperature of a gold

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earring from 25.00C to

30.00C? The earring weighs 25 grams, and the specific heat of gold is 0.128 J/ge0C. 1.

2. 3. Read and

Understand What

information are you

given? Specific heat =

$c = 0.128 \text{ J/ g. } ^\circ\text{C}$  Mass

=  $m = 25.0 \text{ grams}$

### Quia

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Energy And Heat;

Morgan A. • 33 cards.

Heat. the transfer of



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thermal energy from one object to another as the result of a difference in temperature. True. T/F: On the Celsius Scale, the reference points for temperature are the freezing and boiling points of water. thermal energy ...

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